STRUCTURES OF GIBBERELLINS A₂₆ AND A₂₇ IN IMMATURE SEEDS OF <u>PHARBITIS NIL</u> Nobutaka Takahashi, Takao Yokota, Noboru Murofushi and Saburo Tamura Department of Agricultural Chemistry, The University of Tokyo, Bunkyo-ku Tokyo, Japan

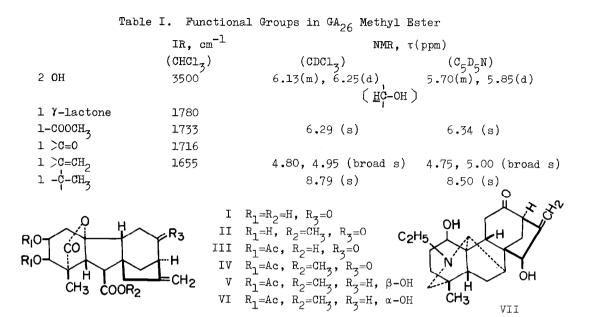
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Isolation and biological activities of new gibberellins A_{26} and A_{27} in immature seeds of the Japanese morning-glory (<u>Pharbitis nil</u>) have been reported (1). We wish to report here the elucidation of their structures.

Gibberellin A_{26} (GA₂₆), m.p. 254-257°, formed a monomethyl ester (II), m.p. 207-211°. The high resolution mass spectrum of II reveales a parent ion peak at m/e 376.159 with the composition $C_{20}H_{24}O_7$ (calcd. 376.152). Therefore GA₂₆ has the molecular formula $C_{19}H_{22}O_7$. The significant peaks at M-32 (CH₄O) and M-60 ($C_2H_4O_2$) in the mass spectrum of II, together with an AB quartet due to the C-10, C-10a protons centered at τ 7.27 and τ 6.63 (J=10cps) in its NMR spectrum (CDCl₃), suggest that GA₂₆ retains the structural features common to the known gibberellins.

As summarized in Table I, the functional groups of II are deduced on the basis of the IR, UV and NMR spectra. GA_{26} rapidly consumed periodate and easily formed a diacetate (III), m.p. 209-210°, having no free hydroxyl group. These suggest the presence of a <u>cis</u>-1,2-glycol system in GA_{26} . An axial hydroxyl at C-2 is indicated by the down-field shifts of the C-1 methyl singlet (τ 8.79 in CDCl₃; τ 8.50 in C₅D₅N) and the C-10a proton doublet (τ 6.63 in CDCl₃; 6.09 in C₅D₅N) in the change of the solvent from CDCl₃ to C₅D₅N (2). Therefore the other hydroxyl must be attached to C-3 in the equatorial conformation. The C-2 proton doublet at τ 6.25 (J=4.0 cps) rationalizes the existence of a hydroxyl substituent at C-3 and the equatorial conformation of the C-3 hydroxyl is supported by the C-3 proton multiplet with a large width (20 cps) characteristic of an axial proton. The pattern of $\underline{H} - \underline{C} - 0H$ region is very similar to that of GA₈. Therefore GA_{26} contains a C-2a, C-3a-glycol.

2077



II shows a UV absorption at λ_{\max} 288 mµ (ϵ 280) and a CD curve having a negative maximum at λ 294 mµ (Θ -8.97 x 10³), which are characteristic of β , γ -unsaturated ketones in the appropriate molecular geometry (3). Since GA₂₆ is considered to retain the structural features common to gibberellins, the position of the ketone should be limited either at C-6 or C-11. The IR band at 1716 cm⁻¹ (CHCl₃) is indicative of a six-membered ring ketone. NaBH₄ reduction of a diacetate of II (IV) afforded an amorphous isomeric mixture of V and VI (4:1), which could be separated on TLC. The NMR spectra of V and VI show new multiplets^{*} of the type $\underline{H} - \dot{\underline{C}}$ -OH at τ 5.88 (W=23 cps) and τ 6.25, respectively. This establishes the position of the ketone at C-6, since an alcohol from a C-11 ketone is expected to give a C-11 proton doublet.

An unusual broad doublet at τ 6.77 (W=10 cps) in the NMR spectrum of IV can be assigned to the C-7 proton, which shifts to the up-field (τ 7.16, W=18 cps) in that of V.

All evidences above cited as well as the consideration of the structural

^{*} These τ values showed that V has an equatorial C-ll proton and VI has an axial C-ll proton because an equatorial proton resonates at lower field than an axial one.

relationship to other members of the gibberellin group allow to assign the structure I for GA_{26} .

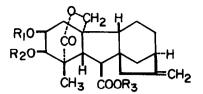
I is further supported by the following observation. An aconite alkaloid, songorine (VII) (4), whose structure has been decided as VII by X-ray analysis, possesses the same structure and stereochemistry of the C,D rings as GA_{26} . The CD curve of VII (a negative maximum at 298 mµ, Θ -1.05 x 10⁴) is quite similar to that of GA_{26} . This furnishes the strong support not only for the position of the ketone but also the stereochemistry of the C,D rings.

 GA_{27} , m.p. 163-165° (hydrate), was converted to a monomethyl ester (IX), m.p. 181-183°. The molecular formula $C_{20}H_{26}O_6$ for GA_{27} was established by the high resolution mass spectrum of IX (a parent ion peak at m/e 376.186; calcd. for $C_{21}H_{28}O_6$, 376.189). Assignments of the NMR and IR spectra of IX for functional groups are illustrated in Table II. The presence of these functional groups in IX as well as the mass spectral features [M-32 (CH_4O) and M-60 ($C_2H_4O_2$)] shows GA_{27} to be a new C_{20} gibberellin.

 GA_{27} rapidly consumed periodate. It readily formed a diacetate (X), m.p. 226-228°, and an isopropylidene derivative (XI), m.p. 216-217°, both of which lack free hydroxyls. These suggest the presence of a <u>cis</u>-1,2-glycol system in GA_{27} . In the similar manner to that for GA_{26} , GA_{27} is shown to have a C-2<u>a</u>,C-3<u>e</u>-glycol.

The presence of a δ -lactone is indicated by the IR band at 1740 or 1710 cm⁻¹ (nujol) and an AB quartet centered at τ 5.54 and 5.88 (J=13 cps) in the NMR spectrum (CDCl₃). The similar NMR patterns have been reported for GA₁₅ (5) and 2,7-dihydroxy-GA₁₅ methyl esters (6) containing δ -lactones between C-1 and C-4a.

Table II. Functional Groups in GA ₂₇ Methyl Ester				
	IR, cm^{-1} (n	ujol) NMR, τ (p)	NMR, τ (ppm)	
		CDC13	C ₅ D ₅ N	
2 OH	3300	6.14 (m), 6.28 (d)	5.65 (m), 5.82 (d)	
		(<u>н</u> -с-он)		
1 -COOCH ₂ -		5.54, 5.88 (AB q)	5.42, 5.81 (AB q)	
$(\delta - lactone)$	1740, 1710	(-COO-C <u>H</u> ₂ -)		
1 -COOCH3		6.31 (s)	6.37 (s)	
l >C=CH2	1650	5.05, 5.17 (broad s)	5.07, 5.21 (broad s)	
1 >C=CH ₂ 1 -¢-CH ₃		8.76 (s)	8.31	



VIII $R_1 = R_2 = R_3 = H$ IX $R_1 = R_2 = H$, $R_3 = CH_3$ X $R_1 = R_2 = Ac$, $R_3 = CH_3$ XI R_1 , $R_2 = (CH_3)_2 C$, $R_3 = CH_3$

A signal due to the C-10, C-10a protons, characteristic of the gibbane skelton, is found as a 2H singlet at τ 7.24 in the NMR spectrum (CDCl₃) of IX, while in C₅D₅N it is splitted into an AB quartet centered at τ 6.73 and 6.96 (J=13 cps). The similar NMR patterns have been observed in the spectra of GA₂₂ (7) and 2,7-dihydroxy-GA₁₅ methyl esters.

The structure VIII for GA_{27} accomodates all evidences above cited.

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